AGROSTISTACHIN, A NOVEL CYTOTOXIC MACROCYCLIC DITERPENE FROM AGROSTISTACHYS HOOKERI

Y.-H. Choi, J. Kim, J.M. Pezzuto, A.D. Kinghorn*, and N.R. Farnsworth Program for Collaborative Research in the Pharmaceutical Sciences College of Pharmacy, University of Illinois at Chicago Chicago, IL 60612, U.S.A.

and H. Lotter* and H. Wagner Institut für Pharmazeutische Biologie der Universität München D-8000 München, W. Germany

Summary: Agrostistachin, a cytotoxic constituent of <u>Agrostistachys</u> <u>hookeri</u>, was characterized as a new member of the rare casbane class of diterpenoids. The structure and stereochemistry of this compound were established by analysis of its spectral and X-ray crystallographic parameters.

In the course of our continuing search for plant-derived antineoplastic agents¹, it was found that a chloroform extract of the twigs of <u>Agrostistachys</u> hookeri Benth. & Hook. f. (Euphorbiaceae) displayed significant in <u>vitro</u> activity against the P-388 lymphocytic leukemia test system, when tested according to standard protocols². Bioactivity-guided fractionation of this extract has led to the isolation of a new crystalline (mp 155-157°, from ethyl acetate) and optically active ($[\alpha]_D + 248°, c$ 0.18, chloroform) constituent, agrostistachin (1), with demonstrated cytotoxicity (ED₅₀, 1.4 µg/ml).



1 R = H $2 R = COCH_3$

By high-resolution mass spectrometry, the elemental formula of the molecular ion of 1 was determined as $C_{20}H_{30}O_3^{-3}$. Resonances in the IR spectrum at 3460, 3396, and 1660 cm⁻¹ suggested that the oxygen-containing functionalities of this molecule were, respectively, a non-hydrogen-bonded hydroxy group, a hydrogen-bonded OH, and an $\alpha\beta$ -unsaturated carbonyl. One olefinic proton appeared as a doublet at δ 6.25 (J = 10 Hz) in the ¹H-NMR spectrum of 1⁴, and was assigned as a proton β to a carbonyl group⁵. A maximum in the UV spectra of 1 occurred at 272.5 nm (EtOH, log ϵ 4.10), a value consistent with this enone moiety being affixed to a cyclopropyl ring^{5,6}. Evidence for the presence of gem-dimethyl functionality in the <u>cis</u>-substituted cyclopropyl ring of 1 was obtained by the observation of signals at δ 1.14 and 0.96 in the ¹H-NMR spectrum, and by the chemical shifts at δ 28.99 and 16.03^{6,7} in its ¹³C-NMR spectrum⁸. The other three methyl groups in the molecule of 1 were vinylic [¹H-NMR, δ 1.87, 1.76, 1.70; ¹³C-NMR (SFORD), singlets at δ 133.41, 137.92, 137.31]. Of the remaining downfield signals in the ¹H-NMR spectrum of 1, two protons that resonated at δ 5.11 were olefinic, while the methine protons at δ 5.22 and 4.18 adjacent to hydroxyl groups were each shifted downfield by about 1 ppm (to δ 6.24 and 5.29, respectively) in the diacetate, 2⁹, produced from 1 by treatment with equal parts of acetic anhydride and pyridine at room temperature.

Compound 1 was tentatively assigned a casbane carbon skeleton¹⁰ by a consideration of the isoprene rule and the allylic coupling evident in its homonuclear ${}^{1}H^{-1}H^{-1}COSY$ 2D-NMR spectrum. An unambiguous assignment of the ${}^{1}H^{-}$ and ${}^{13}C^{-}NMR$ spectra of 1 was provided after the performance of a heteronuclear ${}^{1}H^{-13}C$ chemical shift-correlated 2D-NMR spectral study. The geometry of the three double bonds in the molecule of 1 between position 2,3, position 6,7, and position 12,13, was postulated as <u>E</u>, <u>E</u>, and <u>Z</u>, respectively, on analysis of the ${}^{13}C^{-}NMR$ chemical shifts of the 20- (δ 11.57), 19- (δ 17.83), and 18- (δ 22.89) methyl carbons, as well as the 11-methylene carbon (δ 31.30)^{8,11,12}.

Confirmation of the structure of agrostistachin as 1^{13} was obtained by the application of X-ray crystallography (Fig. 1). Suitable crystals were obtained from diisopropyl ether as transparent colorless needles (0.2 X 0.2 X 1.0 mm), and gave the following crystal data: a = 10.885 Å, b = 12.009 Å, c = 15.219 Å, orthorhombic space group $P2_12_12_1$, Z = 4, $D_{calc.} = 1.06 \text{ gcm}^{-3}$, $D_{meas.} = 1.07 \text{ gcm}^{-3}$. A total of 1566 unique reflections [(1453 observed I ≥ 30 (I)] were measured on a Nicolet T3m diffractometer with Ni-filtered CuK_{α}-radiation. The structure was solved by direct methods using SHELXTL¹⁴. The first attempt showed all nonhydrogen atoms in an E-map, while the refinement with unit weights converged at R = 9.7% (without hydrogen). Atomic coordinates and bond lengths are deposited at the Cambridge Crystallographic Data Centre.



Figure 1. Crystal structure of agrostistachin (1).

Agrostistachin (1) represents the first casbane-type diterpenoid to exhibit cytotoxic activity. The parent compound, (-)-casbene, which has a 2,6,12-all-<u>E</u>-triene system, is regarded as an important biogenetic precursor of biologically active cyclopropanoid diterpenes of the phorbol ester type, some of which are widely used in chemical carcinogenesis experiments^{6,15}. While agrostistachin (1) and the monoolefin, crotonitenone^{5,16}, are the only oxygenated casbane derivatives so far known, 1 may be regarded as being more biogenetically primitive than the latter compound, in having three double bonds in the same positions as (-)-casbene. However, agrostistachin (1) is the first casbane diterpene with Δ^{12} -cis-stereochemistry, and the C-14 hydroxy group represents an unusual oxidation pattern of potential importance in the study of the biosynthesis of macrocyclic diterpenoids of the Euphorbia-ceae.

Acknowledgment: This work was supported by grant CA-33047 with the National Cancer Institute, National Institutes of Health, Bethesda, Maryland. The authors thank the Research Resources Center of the University of Illinois at Chicago for the provision of spectroscopic facilities, and Mrs. M. Sitt for help in the preparation of this manuscript. JMP is the recipient of a Research Career Development Award from the National Cancer Institute, 1984–1989.

References and Notes:

- 1. Part XLV in the series "Plant Anticancer Agents". For the previous paper in this series, see: Duh, C.-Y.; Pezzuto, J.M.; Kinghorn, A.D.; Leung, S.L.; Farnsworth, N.R. J. Nat. Prod., submitted.
- 2. Arisawa, M.; Pezzuto, J.M.; Bevelle, C.; Cordell, G.A. J. Nat. Prod. 1984, 47, 453.
- 3. MS of 1: Mass measurement, M^{+} , $\underline{m/z}$ 318.2198, calc. for $C_{20}H_{30}O_3$, 318.2195; CI (CH₄), $\underline{m/z}$ 319 (M^{+} + H, 30%), 301 (M^{+} OH, 100).
- 4. ¹H-NMR of 1: δ (360 MHz, CDCl₃) 6.25 (1H, d, $\underline{J}_{7,8} = 10$ Hz, 7-H), 5.22 (1H, d, $\underline{J}_{3,4} = 10$ Hz, 4-H), 5.11 (2H, br. d, 3-H, 13-H), 4.18 (1H, m, 14-H), 2.44 (1H, dd, $\underline{J}_{gem} = 12$ Hz, $\underline{J}_{1,14} = 2.5$ Hz, 1-H), 2.25 (1H, m, 11-H), 2.01 (2H, m, 10-H, 1-H), 1.87 (3H, d, $\underline{J}_{7,20} = 0.7$ Hz, 20-CH₃), 1.78 (1H, m, 11-H), 1.76 (3H, d, $\underline{J}_{3,19} = 1.5$ Hz, 19-CH₃), 1.70 (3H, br. s, 18-CH₃), 1.51 (1H, dd, $\underline{J}_{7,8} = 10$ Hz, $\underline{J}_{8,9} = 8$ Hz, 8-H), 1.14 (3H, s, 17-CH₃), 1.11 (1H, m, 9-H), 0.96 (3H, s, 16-CH₃), 0.83 (1H, m, 10-H).
- Burke, B.A.; Chan, W.R.; Pascoe, K.O.; Blount, J.F.; Manchand, P.S. J. Chem. Soc., Perkin I, 1981, 2666.
- 6. Edgar, H.S.; Hecker, E. Phytochemistry, 1983, 22, 1791.
- 7. Crombie, L.; Kneen, G.; Pattenden, G.; Whybrow, D. J. Chem. Soc., Perkin I, 1980, 1711.
- ¹³C-NMR spectrum of 1: δ (90.8 MHz, CDCl₃) 199.02 (s, 5-C=O), 145.06 (d, 7-C), 137.92 (s, 2-C), 137.31 (s, 12-C), 133.41 (s, 6-C), 128.67 (d, 3-C), 127.08 (d, 13-C), 70.21 (d, 14-C), 68.32 (d, 4-C), 47.56 (t, 1-C), 34.57 (d, 9-C), 31.70 (t, 11-C), 28.99 (q, 17-C), 27.47 (d, 8-C), 27.31 (s, 15-C), 22.89 (t, 10-C), 22.83 (q, 18-C), 17.83 (q, 19-C), 16.03 (q, 16-C), 11.57 (q, 20-C).

- 9. The diacetate, **2**, exhibited the following data: mp 115-117 ${}^{\circ}$ C; $[\Box]_{D} + 203^{\circ}$ (<u>c</u> 0.12, chloroform); UV, (EtOH) λ_{max} 272.5 nm (log \in 4.06); IR, ν_{max} 1733, 1677, 1625 cm⁻¹; ¹H-NMR, δ (360 MHz, CDCl₃), 6.24 (2H, br. d, 4-H, 7-H), 5.40 (1H, d, <u>J</u>_{3,4} = 10 Hz, 3-H), 5.29 (1H, m, 14-H), 5.08 (1H, d, <u>J</u>_{13,14} = 10 Hz, 13-H), 2.46 (1H, dd, <u>J</u>_{gem} = 12 Hz, <u>J</u>_{1,14} = 3 Hz, 1-H), 2.44 (1H, m, 11-H), 2.15 (1H, m, 1-H), 2.14 (3H, s, -OAe), 2.10 (1H, m, 10-H), 1.99 (3H, s, -OAe), 1.90 (3H, br. s, 20-CH₃), 1.83 (1H, m, 11-H), 1.78 (3H, d, <u>J</u>_{3,19} = 1.3 Hz, 19-CH₃), 1.76 (3H, br. s, 18-CH₃), 1.54 (1H, dd, <u>J</u>_{7,8} = 10 Hz, <u>J</u>_{8,9} = 8 Hz, 8-H), 1.18 (3H, s, 17-CH₃), 1.13 (1H, m, 9-H), 1.04 (3H, s, 16-CH₃), 0.91 (1H, m, 10-H); ¹³C-NMR, δ (90.8 MHz, CDCl₃) 199.21 (s, 5-C), 170.22 (s, <u>COCH₃</u>), 169.76 (s, -COCH₃), 142.73 (d, 7-C), 140.89 (s, 2-C), 140.35 (s, 12-C), 134.89 (s, 6-C), 124.02 (d, 3-C), 123.34 (d, 13-C), 72.53 (d, 14-C), 71.39 (d, 4-C), 44.36 (t, 1-C), 22.81 (q, 18-C), 21.26 (q, -CO<u>CH₃</u>), 20.72 (q, -CO<u>CH₃</u>), 18.58 (q, 17-C), 16.14 (q, 17-C), 11.61 (q, 20-CH₃); MS, CI (CH₄) m/z 403 (M⁺ + 1, 4%), 343 (M⁺ + H OAc, 72.7), 283 (M⁺ + H 2 x OAc, 100).
- 10. The numbering system adopted for agrostistachin (1) and its acetate (2) is that published for the oxygenated casbane derivative, crotonitenone (see ref. 5).
- 11. Barlow, L.; Pattenden, G. J. Chem. Soc., Perkin I, 1976, 1029.
- 12. Wiemer, D.F.; Meinwald, J.; Prestwich, G.D.; Miura, I. J. Org. Chem. 1979, 44, 3950.
- 13. The systematic nomenclature for 1 based on the <u>Chemical Abstracts</u> system is therefore {(E,E,Z)-(1S*,5R*,9R*,14R*)-5,9-dihydroxy-3,7,11,15,15-pentamethylbicyclo[12.1.0]pentadec-2,6,10-trien-4-one}-(+).
- 14. Sheldrick, G.M. SHELXTL (Release 4.1), A Program for Crystal Structure Determination, Cambridge/Gottingen, 1983.
- 15. Adolf, W.; Hecker, E. Isr. J. Chem. 1977, 16, 75.
- 16. Commissiong, M.A.; Pascoe, K. Tetrahedron Lett. 1984, 25, 711.

(Received in USA 16 July 1986)